

Unit 11: Chapter 7 - Chemical Reactions Review Answer Key

Define:

1. **Reactant:** *A substance that undergoes change in a chemical reaction.*
2. **Product:** *New substances formed as a result of a chemical reaction.*
3. **Catalyst:** *A substance that affects the rate of a chemical reaction without being used up in the reaction.*
4. **Coefficient:** *Numbers that appear before a formula in a chemical equation to show the relative proportions of each reactant and product.*
5. **Subscript:** *A number written lower in a chemical formula indicating the amount of atoms present of an element (this number is not to be changed once written in chemical formula).*
6. **Law of Conservation of Mass:** *Matter cannot be created nor destroyed; only changed (atoms can be rearranged).*
7. **Synthesis reaction (include an example):** *A chemical reaction in which two or more substances react to form a single substance.* $(A + B \rightarrow AB)$
8. **Decomposition reaction (include an example):** *A chemical reaction in which a compound breaks down into two or more simpler substances.* $(AB \rightarrow A + B)$
9. **Single replacement (include an example):** *A chemical reaction in which one element takes the place of another element in a compound.* $(AB + C \rightarrow AC + B)$
10. **Double replacement (include an example):** *A chemical reaction in which two compounds exchange positive ions and form two new compounds.* $(AB + CD \rightarrow AD + CB)$
11. **Combustion (include an example):** *A chemical reaction in which a substance reacts rapidly with oxygen, often producing heat and light.* $(C_2H_6 + O_2 \rightarrow CO_2 + H_2O)$
12. **Endothermic reaction:** *A chemical reaction that absorbs energy from its surroundings.*
13. **Exothermic reaction:** *A chemical reaction that releases energy to its surroundings.*
14. **Chemical equation (include an example):** *A representation of a chemical reaction in which the reactants and products are expressed as formulas.*
15. **A balanced equation** is an equation in which the number of atoms on the reactants' side equals the number of atoms on the products' side of the chemical equation.
16. **Equilibrium:** *A state in which the forward and reverse paths of a physical or chemical change take place at the same rate.*

Reactions Review Answer

Minjie Lin

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Reactions Review Answer:

Decoding **Reactions Review Answer**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Reactions Review Answer**," a mesmerizing literary creation penned by a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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Table of Contents **Reactions Review Answer**

1. Understanding the eBook **Reactions Review Answer**
 - The Rise of Digital Reading **Reactions Review Answer**
 - Advantages of eBooks Over Traditional Books
2. Identifying **Reactions Review Answer**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an **Reactions Review Answer**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **Reactions Review Answer**
 - Personalized Recommendations
 - **Reactions Review Answer** User Reviews and Ratings
 - **Reactions Review Answer** and Bestseller Lists

5. Accessing Reactions Review Answer Free and Paid eBooks
 - Reactions Review Answer Public Domain eBooks
 - Reactions Review Answer eBook Subscription Services
 - Reactions Review Answer Budget-Friendly Options
6. Navigating Reactions Review Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Reactions Review Answer Compatibility with Devices
 - Reactions Review Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reactions Review Answer
 - Highlighting and Note-Taking Reactions Review Answer
 - Interactive Elements Reactions Review Answer
8. Staying Engaged with Reactions Review Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reactions Review Answer
9. Balancing eBooks and Physical Books Reactions Review Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reactions Review Answer
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Reactions Review Answer
 - Setting Reading Goals Reactions Review Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Reactions Review Answer
 - Fact-Checking eBook Content of Reactions Review Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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