

CHAPTER 13

Rates of Reaction

■ SOLUTIONS TO EXERCISES

Acetaminophen is a painkiller that gives analgesic results in the treatment of headaches with less dependence upon the use of other painkillers in combination. The high amount of the compound, the most common painkiller, is responsible for the dangerous side effects associated with its use. The compound is a white crystalline solid that is soluble in water, but not in alcohol. It is a weak acid and is a weak base.

- 13.1. The rate of reaction is $0.0012 \text{ mol/L} \cdot \text{s}$. The rate of reaction is $0.0012 \text{ mol/L} \cdot \text{s}$. The rate of reaction is $0.0012 \text{ mol/L} \cdot \text{s}$. The rate of reaction is $0.0012 \text{ mol/L} \cdot \text{s}$.

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