

PHYS 202 DRILL HOMEWORK PROBLEMS

PART 4

- D4-1. A metal object is heated to 200°C . At wavelength is most of its thermal radiation emitted? In what portion of the spectrum is this light?
[6.1 μm , infrared]
- D4-2. The sun's peak emission is around 580 nm. Estimate the surface temperature of the sun.
[5000 K]
- D4-3. A cube whose sides are 3 cm long is at a temperature of 80°C . Assuming that its emissivity is one, find the total power it emits in thermal radiation.
[4.76 W]
- D4-4. Suppose the cube in the previous problem is inside a box whose walls are at 30°C . Find the rate at which the cube absorbs thermal energy emitted by the walls of the box.
[2.59 W]
- D4-5. A green laser emits 0.01 W of power at a wavelength of 514 nm. Find the energy of a single photon and the rate that the laser emits the photons.
[2.418 eV = 3.87×10^{-19} J, 2.58×10^{16} per second]
- D4-6. Find the wavelengths of the photons with the following energies: 0.1 eV, 2 eV, 6 eV. Specify the portion of the spectrum that each photon is in.
[12.4 μm (IR), 622 nm (vis), 207 nm (uv)]
- D4-7. Which photon carries more energy, a blue photon or a red photon? If a blue light and a red light emit the same optical power, which one emits more photons each second?
[blue; red]
- D4-8. A certain metal has a work function of 1.95 eV. What is the longest wavelength that a photon can have if it is to be absorbed and emit an electron from the metal?
[637 nm]
- D4-9. A photon with a wavelength of 615 nm hits a metal with a work function of 2.2 eV. Can an electron be ejected? If so, find the maximum kinetic energy that it can have as it leaves.
[No]
- D4-10. A photon with a wavelength of 535 nm hits a metal with a work function of 2.2 eV. Can an electron be ejected? If so, find the maximum kinetic energy that it can have as it leaves.
[Yes, 0.12 eV]
- D4-11. A Hydrogen atom makes a transition from the $n=4$ state to the $n=2$ state. It then makes another transition to the ground state. Find the wavelengths of the two emitted photons and specify their types.
[4 $\rightarrow$ 2: 487.5 nm (blue), 2 $\rightarrow$ 1: 121.9 nm (uv)]

Physics Supplemental Problems

Robert R. Redfield



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