

After this time the activity of original sample: (1.2 marks)
 10000 gives 2.0 Bq
 Volume of water in system: (1.2 marks)
 (1.2 marks)

5. A sample of phosphorus-32 has an activity of 1.6×10^5 Bq. Phosphorus-32 has a half-life of 14.3 days.

- a. Calculate the decay constant.

$$(1.2 \text{ marks})$$

$$\lambda = 0.0485 \text{ days}^{-1} \quad (1.2 \text{ marks})$$

$$(1.2 \text{ marks})$$

- b. Calculate time taken for the activity to fall to 1000 Bq.

$$A = A_0 e^{-\lambda t}$$

$$1000 = 1.6 \times 10^5 e^{-\lambda t} \quad (1.2 \text{ marks})$$

$$0.3875 = e^{-\lambda t}$$

$$\ln 0.3875 = -\lambda t \quad (1.2 \text{ marks})$$

$$-0.9474 = -0.0485 t$$

$$t = 19.5 \text{ days} \quad (1.2 \text{ marks})$$

$$(1.2 \text{ marks})$$

6. A sample of iodine-131 which has a half-life of 8.04 days has an initial activity of 7.4×10^7 Bq. Calculate the activity after 20 days.

$$(1.2 \text{ marks})$$

$$\lambda = 0.0863 \text{ days}^{-1} \quad (1.2 \text{ marks})$$

$$A = A_0 e^{-\lambda t}$$

$$A = 7.4 \times 10^7 e^{-\lambda t} \quad (1.2 \text{ marks})$$

$$= 6.642 \times 10^7 \text{ Bq}$$

$$= 6.6 \times 10^7 \text{ Bq} \quad (1.2 \text{ marks})$$

$$(1.2 \text{ marks})$$

7. A radioactive source has an initial count rate of 133 counts per second. After 10 minutes the count rate has dropped to 54 counts per second.

Background radiation is 8 counts per second.

- a. Calculate the decay constant.

Subtract background from each activity

$$A = A_0 e^{-\lambda t}$$

$$92 = 125 e^{-\lambda t} \quad (1.2 \text{ marks})$$

$$\ln 0.736 = -\lambda t \quad (1.2 \text{ marks})$$

$$-0.308 = -\lambda \times 600$$

$$\lambda = 5.1 \times 10^{-4} \text{ s}^{-1} \quad (1.2 \text{ marks})$$

$$(1.2 \text{ marks})$$

- b. Calculate the number emitted in the original sample if it can be assumed that the detector counts all of the radiation emitted from the source.

$$A = \lambda N$$

$$125 = 5.1 \times 10^{-4} N \quad (1.2 \text{ marks})$$

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Y Pai



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