

Ideal Gas Law

The ideal gas law is the equation of state of an ideal gas that relates pressure, volume, quantity of gas, and absolute temperature.

$$PV = nRT$$

P = pressure

V = volume

n = number of moles of gas

R = ideal gas constant

T = absolute temperature

Although it applies to ideal gases, it approximates the behavior of many real gases.

Physics Ideal Gas Law Labs Answers

Thomas Bond, Chris Hughes



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