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p = m + v - ANS-density equation different states of matter, differences in density - ANS-particle model explains substances change state - ANS-mass is conserved when physical - ANS-a change of state is no new substance is produced - ANS-during a change of state reversed - ANS-a substance recovers its original properties if the change is electrons=protons - ANS-atoms have no overall charge because same number of protons - ANS-atoms of the same element have different numbers of protons - ANS-atoms of different elements have atoms of the same element with different numbers of neutrons - ANS-isotopes are different energy level - ANS-atoms are arranged in shells, each at a 0.1nm (1x10⁻¹⁰) - ANS-approximate radius of an atom positive - ANS-ions: atoms lose one or more electrons negative - ANS-ions: atoms gain one or more electrons nucleus - ANS-most of an atom's mass is in the 1/10000 of an atom (1x10⁻¹⁴) - ANS-radius of nucleus higher energy level - ANS-EM absorption causes electrons to move to a lower energy level - ANS-EM emission causes electrons to move to a protons+neutrons - ANS-mass number= protons - ANS-atomic number= gold foil - ANS-a scattering; fired thin a particles into straight through detector - ANS-a scattering: most went deflected - ANS-a scattering: some

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