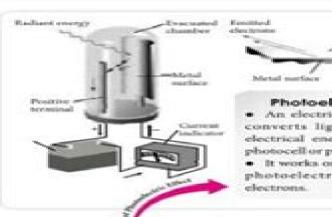
QUANTUM THEORY OF LIGHT





Photoelectric Cell

- · An electrical device which converts light energy into electrical energy, is called as photocell or photoelectric cell.
- · It works on the principle of photoelectric emission of

Photoelectric Effect

· The phenomenon of emission of electrons from a metal surface when an electromagnetic wave of suitable frequency is incident on it is called photoelectric officet.

Basic Quantum Theory of Light

According to Planck, the energy of a photon, E = 0;

$$E = \hbar v_0 = \frac{\hbar c}{\lambda} = \frac{1240}{\lambda (\ln nm)} eV$$

- Momentum of photon, $p = \frac{E}{n} = \frac{hv}{n} = \frac{h}{n}$
- If source is 100% efficient, then the number of photons emitted per second by the source can be given by

$$= \frac{\text{Power of source}}{\text{Energy of photon}} = \frac{P}{E} = \frac{P}{hv} = \frac{P\lambda}{hc}$$

- The energy crossing per unit area per unit time perpendicular to the direction of propagation is called the intensity of a wave. I = E / A E = P / A
- · Force exerted on perfectly reflecting surface

$$F = \frac{\Delta p}{t} = \frac{2Nh}{t\lambda} = n\left(\frac{2h}{\lambda}\right) = \frac{2P}{c}$$

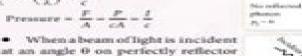
$$F = 2P - 2I$$

Pressure =
$$\frac{F}{A} = \frac{2P}{cA} = \frac{2I}{c}$$

Force exerted on perfectly absorbing surface

$$F = \frac{\Delta p}{t} = \frac{Nh}{t\lambda} = n\left(\frac{h}{\lambda}\right) = \frac{p}{c}$$

Pressure =
$$\frac{F}{A} - \frac{P}{cA} - \frac{I}{c}$$





surface then force exerted on the surface,
$$F = \frac{2P}{c}\cos\theta = \frac{2JA\cos\theta}{c}.$$
 Pressure = $\frac{2J\cos\theta}{c}$

$$F = \frac{2P}{\epsilon} \cos \theta = \frac{2IA \cos \theta}{\epsilon}$$

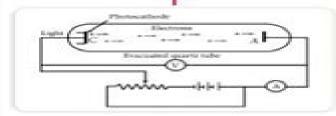
$$Pressure = \frac{2I \cos \theta}{\epsilon}$$

Photoelectric Equation

• $E = K_{max} + \Phi_n$ where & = work function of metal. E = energy of incident light, $K_{max} = maximum kinetic energy of electrons$

Principal Services (Services)

 $=\frac{1}{2}mv_{state}^2 = h(\upsilon - \upsilon_0) = hc\left(\frac{1}{\lambda} - \frac{1}{\lambda_{\infty}}\right)$ where, $\lambda_0 = \frac{hc}{\lambda}$ = threshold wavelength



Conclusions of Experimental Study of Photoelectric Effect

PARTICLE

NATURE OF

RADIATION

Photo-current is directly proportional to the ... intensity of incident light, i.e., $i_n \approx I$.

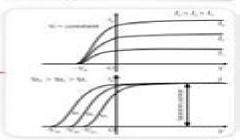
(At constant frequency trand potential V)

- · At constant frequency and intensity, the minimum negative potential at which the photocurrent becomes zero is called stopping protential (V_n) .
- At stopping potential V₀, K_{max} = eV₀
- · For a given frequency of the incident radiation, the stopping potential is independent of its internsity.
- · The stopping potential varies linearly with the frequency of incident radiation but saturation current value remains constant for a fixed intensity of incident radiation.



n, = number of electron emitted persecond

n_{ph} = total number of photon incident per second



Physics Study Guide Quantum Theory

R Sandford

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