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Chapter 1

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Answers to "Why?" in the text

Page 31:

Oxygen has an atomic mass of 16 whereas it is 14 for nitrogen. The O₂ molecule is therefore heavier than the N₂ molecule. Thus, from $\frac{1}{2}m\overline{v^2} = 3(\frac{1}{2}kT)$, the rms velocity of O₂ molecules is smaller than that of N₂ molecules.

Page 34, footnote 11

For small extensions, the difference between the engineering and instantaneous strains due to a temperature change are the same. Historically, mechanical and civil engineers measured extension by monitoring the change in length, ΔL ; and the instantaneous length L was not measured. It is not trivial to measure both the instantaneous length and the extension simultaneously. However, since we know L_v and measure ΔL , the instantons length $L = L_v + \Delta L$. Is the difference important? Consider a sample of length L_v that extends to a final length L due to a temperature change from T_v to T. Let $\varepsilon = (L - L_v) / L_v = \Delta L / L_v$ be the engineering strain.

The engineering definition of strain and hence the thermal expansion coefficient is

Engineering strain =
$$\frac{\delta L}{L_a} = \lambda \delta T$$

so that thermal expansion from T_o to T gives,

$$\int_{L}^{L} \frac{dL}{L_{o}} = \int_{T}^{T} \lambda dT \qquad \therefore \qquad \frac{\Delta L}{L_{o}} = \lambda (T - T_{o}) \qquad \therefore \qquad \varepsilon = \lambda (T - T_{o}) \qquad (1)$$

where $\varepsilon = \Delta L/L_o$ is the engineering strain as defined above

Physics definition of strain and hence the thermal expansion coefficient is

Instantaneous train =
$$\frac{\delta L}{L} = \lambda \delta T$$

so that thermal expansion from To to T gives,

$$\int_{L}^{L} \frac{dL}{L} = \int_{T_{c}}^{T} \lambda dT \qquad \therefore \qquad \ln\left(\frac{L}{L_{o}}\right) = \lambda(T - T_{o}) \qquad \therefore \qquad \ln\left(1 + \varepsilon\right) = \lambda(T - T_{o}) \qquad (2)$$

We can expand the $ln(1 + \epsilon)$ term for small ϵ , so that Equation (2) essentially becomes Equation (1)

1.1 Virial theorem The Li atom has a nucleus with a +3e positive charge, which is surrounded by a full 1s shell with two electrons, and a single valence electron in the outer 2s subshell. The atomic radius of the

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Puja Mehta

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a 400 would have been yanked within days of showing up in the yard and the 350 is both the base engine chrysler used this engine family from the 1920s through the 1970s

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