

System - an object or a group of objects
When a system changes, the way energy is stored also changes.

EXAMPLES:

Ball Rolling and hitting a wall:

- System is the moving ball
- When it hits the wall, some of the kinetic energy is transferred as sound

Vehicle Slowing Down:

- System is moving vehicle
- When it slows down, kinetic energy transfers to thermal due to friction between wheels and brakes

Specific Heat Capacity - the energy required to raise the temperature of 1kg of a substance by 1°C

Energy = mass x specific heat capacity x temperature change

$$E = mc \Delta T$$

E = energy (J)

m = mass (kg)

c = specific heat capacity (J/kg°C)

T = temperature (°C)

Kinetic Energy

$$E_k = \frac{1}{2} m v^2$$

m = mass (kg)

v = speed (m/s)

E_k = kinetic energy (J)

Elastic Potential Energy

$$E_e = \frac{1}{2} k e^2$$

k = spring constant (N/m)

e = extension (m)

E_e = elastic potential energy (J)

Gravitational Potential Energy

$$GPE = mgh$$

GPE = Gravitational Potential Energy

m = mass (kg)

g = gravitational field strength (N/kg)

h = height (m)

Power - the rate at which energy is transferred or the rate at which work is done

$$P = \frac{E}{t} \quad P = \frac{W}{t}$$

P = Power (W)

E = Energy (J)

W = Work Done (J)

t = time (s)

An energy transfer of 1 joule per second is equal to a power of 1 watt.

If given two motors A and B, the motor that can do the same work faster is more powerful - as the energy is transferred at a faster rate.

Energy Stores

Kinetic	Moving objects have kinetic energy
Thermal	All objects have thermal energy
Chemical	Anything that can release energy during a chemical reaction
Elastic Potential	Things that are stretched
Gravitational Potential	Anything that is raised
Electrostatic	Charges that attract or repel
Magnetic	Magnets that attract or repel
Nuclear	The nucleus of an atom releases energy

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Camilla Rothe



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