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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$ reduction $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ag}^{+} + \text{Ni}$ 3 $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ag}^{+} + \text{Ni}$ 25 $\text{Cu} + \text{Pb}^{2+} \rightarrow \text{Cu}^{2+} + \text{Pb}$ 26 $\text{O}_2 \rightarrow \text{O}^{2-}$

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