Name: ANSWER KEY
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THE PERIODIC TABLE

Review Worksheet

1. Fill in the blanks with the terms and words most appropriate to complete each sentence.

a) In the mid-nineteenth century, only sixty or so elements had been discovered. In 1869, a year's a Russian chemist and inventor named Dmitri Mendeleev created a chart called the periodic table that predicted the properties of elements that had not yet been discovered. He did this by arranging the elements according to their increasing atomic numbers and by their physical and chemical properties.

b) Each element in the periodic table is represented by an element box. In this box there is the following information: two numbers, one is a whole number which is called the atomic number, and the other number is larger and contains a decimal, and this one is called the atomic mass. In the box, there is also the short form of the element called the element symbol and then the full identity of the element which is written using the element name.

c) Today the modern periodic table is arranged into columns and rows. The columns are called groups and there are 18 (a number) of them. The rows are called periods and there are 7 (a number) of them. The periodic table is divided by a zigzag line also known as the staircase that separates the metals found on the left of it, from the non-metals found on the right of it. Directly around the zigzag line are eight elements called metalloids.

d) All the metals, except for mercury which is a liquid, are found as solids at room temperature. The metals are also mostly silvery in color, are ductile, malleable, conduct electricity and some are attracted to magnets.

e) At room temperature, most of the non-metals are found in the gas state except for bromine which is a liquid, and iodine, astatine, carbon, phosphorus, and sulfur which are all found as solids. These elements are colorless or come in various colors, are not ductile, not malleable and do not conduct electricity nor are they attracted to magnets.

f) Metalloids are elements that have some properties of metals and some properties of non-metals. They are all found in the solid state. Most have a silvery metallic luster, they are brittle, and are medium to good electrical conductors.

g) Chemical families are groups on the periodic table that contain elements that share similar properties.

h) Group 1 elements are called the alkali metals and they contain the most reactive group of all the metals. They react very violently with water and therefore must be stored under mineral oil so that the water vapor in the air cannot react with these elements. These metals get more reactive as you go down (up or down) the group and the most reactive of all the elements in this group is francium.

i) Group 2 elements are called the alkaline earth metals and they contain the second most reactive group of all the metals. They also react with water and get more reactive as you go down (up or down) the group.

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Table of Contents Periodic Table Mixed Review Answer Key

1. Understanding the eBook Periodic Table Mixed Review Answer Key
    - The Rise of Digital Reading Periodic Table Mixed Review Answer Key
    - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Mixed Review Answer Key
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Periodic Table Mixed Review Answer Key
    - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Mixed Review Answer Key
    - Personalized Recommendations
    - Periodic Table Mixed Review Answer Key User Reviews and Ratings
    - Periodic Table Mixed Review Answer Key and Bestseller Lists
5. Accessing Periodic Table Mixed Review Answer Key Free and Paid eBooks
    - Periodic Table Mixed Review Answer Key Public Domain eBooks
    - Periodic Table Mixed Review Answer Key eBook Subscription Services
    - Periodic Table Mixed Review Answer Key Budget-Friendly Options

6. Navigating Periodic Table Mixed Review Answer Key eBook Formats
    - ePub, PDF, MOBI, and More
    - Periodic Table Mixed Review Answer Key Compatibility with Devices
    - Periodic Table Mixed Review Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Periodic Table Mixed Review Answer Key
    - Highlighting and Note-Taking Periodic Table Mixed Review Answer Key
    - Interactive Elements Periodic Table Mixed Review Answer Key
8. Staying Engaged with Periodic Table Mixed Review Answer Key
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Periodic Table Mixed Review Answer Key
9. Balancing eBooks and Physical Books Periodic Table Mixed Review Answer Key
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Periodic Table Mixed Review Answer Key
10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Mixed Review Answer Key
    - Setting Reading Goals Periodic Table Mixed Review Answer Key
    - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Mixed Review Answer Key
    - Fact-Checking eBook Content of Periodic Table Mixed Review Answer Key
    - Distinguishing Credible Sources
13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
14. Embracing eBook Trends
    - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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