

PERIODIC TABLE (BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H Hydrogen 1.008	2 He Helium 4.0026																
3 Li Lithium 6.941	4 Be Beryllium 9.0122																
5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180												
11 Na Sodium 22.990	12 Mg Magnesium 24.305	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.948										
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 52.00	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.69	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80
37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.906	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.41	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.757	52 Te Tellurium 127.6	53 I Iodine 126.905	54 Xe Xenon 131.29
55 Cs Cesium 132.905	56 Ba Barium 137.327																
57 Pr Praseodymium 140.908	58 Nd Neodymium 144.24	59 Pm Promethium 144.913	60 Ce Cerium 140.12	61 La Lanthanum 138.905	62 Ce Cerium 140.12	63 Pr Praseodymium 140.908	64 Nd Neodymium 144.24	65 Pm Promethium 144.913	66 Sm Samarium 150.36	67 Eu Europium 151.964	68 Gd Gadolinium 157.25	69 Tb Terbium 158.925	70 Dy Dysprosium 162.50	71 Ho Holmium 164.930	72 Er Erbium 167.259	73 Tm Thulium 168.930	74 Yb Ytterbium 173.054
75 Lu Lutetium 174.967	76 Hf Hafnium 178.49	77 Ta Tantalum 180.948	78 W Tungsten 183.84	79 Re Rhenium 186.207	80 Os Osmium 190.23	81 Ir Iridium 192.222	82 Pt Platinum 195.084	83 Au Gold 196.967	84 Hg Mercury 200.59	85 Tl Thallium 204.38	86 Pb Lead 207.2	87 Bi Bismuth 208.980	88 Po Polonium 209	89 At Astatine 210	90 Rn Radon 222	91 Ac Actinium 227	92 Th Thorium 232.038
93 Pa Protactinium 231.036	94 U Uranium 238.029	95 Np Neptunium 237.048	96 Pu Plutonium 244.064	97 Am Americium 243.061	98 Cm Curium 247.070	99 Bk Berkelium 247.070	100 Es Einsteinium 252.083	101 Fm Fermium 257.105	102 Md Mendelevium 258.10	103 No Nobelium 259.10	104 Lr Lawrencium 262	105 Nh Nihonium 286	106 Ds Darmstadtium 271	107 Ts Tennessine 289	108 Og Oganesson 294	109 Uut Ununtrium 288	110 Uuq Ununquadium 289
111 Rg Roentgenium 281	112 Uub Ununbium 285	113 Uut Ununtrium 284	114 Uuq Ununquadium 289	115 Uup Ununpentium 288	116 Uuh Ununhexium 292	117 Uus Ununseptium 291	118 Uuo Ununoctium 294	119 Uut Ununtrium 293	120 Uuq Ununquadium 297	121 Uup Ununpentium 296	122 Uuh Ununhexium 299	123 Uus Ununseptium 301	124 Uuo Ununoctium 304	125 Uut Ununtrium 303	126 Uuq Ununquadium 307	127 Uup Ununpentium 306	128 Uuh Ununhexium 310

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic forms (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

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An advanced

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Curtis,Marci Smith Deal,Christina Gawlik,Candace Joswick,Melissa Hulings,Kathleen C. Tice,2023-01-03 REA s TExES Core

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1. Understanding the eBook Periodic Table Basic Speedy Study Guides
 - The Rise of Digital Reading Periodic Table Basic Speedy Study Guides
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Basic Speedy Study Guides
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Basic Speedy Study Guides
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Basic Speedy Study Guides
 - Personalized Recommendations
 - Periodic Table Basic Speedy Study Guides User Reviews and Ratings
 - Periodic Table Basic Speedy Study Guides and Bestseller Lists
5. Accessing Periodic Table Basic Speedy Study Guides Free and Paid eBooks
 - Periodic Table Basic Speedy Study Guides Public Domain eBooks
 - Periodic Table Basic Speedy Study Guides eBook Subscription Services
 - Periodic Table Basic Speedy Study Guides Budget-Friendly Options

6. Navigating Periodic Table Basic Speedy Study Guides eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Basic Speedy Study Guides Compatibility with Devices
 - Periodic Table Basic Speedy Study Guides Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Basic Speedy Study Guides
 - Highlighting and Note-Taking Periodic Table Basic Speedy Study Guides
 - Interactive Elements Periodic Table Basic Speedy Study Guides
8. Staying Engaged with Periodic Table Basic Speedy Study Guides
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Basic Speedy Study Guides
9. Balancing eBooks and Physical Books Periodic Table Basic Speedy Study Guides
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Basic Speedy Study Guides
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Basic Speedy Study Guides
 - Setting Reading Goals Periodic Table Basic Speedy Study Guides
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Basic Speedy Study Guides
 - Fact-Checking eBook Content of Periodic Table Basic Speedy Study Guides
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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- Interactive and Gamified eBooks

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