










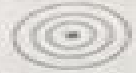
















Model 1 – Main Group Elements

1 H 								2 He 
37 1312 2.1								31 2372 N/A
3 Li 	4 Be 	5 B 	6 C 	7 N 	8 O 	9 F 	10 Ne 	
152	112	83	77	71	66	71	70	
520	900	801	1086	1402	1314	1681	2081	
1.0	1.5	2.0	2.5	3.0	3.5	4.0	N/A	
11 Na 	12 Mg 	13 Al 	14 Si 	15 P 	16 S 	17 Cl 	18 Ar 	
186	160	143	117	115	104	99	98	
496	738	578	786	1011	1000	1251	1521	
0.9	1.2	1.5	1.8	2.1	2.5	3.0	N/A	
19 K 	20 Ca 	31 Ga 	32 Ge 	33 As 	34 Se 	35 Br 	36 Kr 	
227	197	122	123	125	117	114	112	
404	550	558	709	834	869	1008	1170	
0.8	1.0	1.7	1.8	1.9	2.1	2.5	N/A	

Atomic Number
Element Symbol
Electron Shell Diagram
Atomic Radius (pm)
1st Ionization Energy (kJ/mol)
Electronegativity

Note: The transition elements and f-block elements have been removed from the periodic table here to ease the analysis of the trends.

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JS Bruner



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Chicago, Ill. October 25, 1961, pt. 2 Los Angeles, Calif. November 15 and 16, 1961, pt. 3 New York, N.Y. November 29 and 30, 1961, pt. 4 Washington, D.C., May 1, 2, and 4, 1962 United States. Congress. House. Committee on Post Office and Civil Service. Subcommittee on Census and Statistics, 1962

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