
The Periodic Law

Section Quiz: History of the Periodic Table

As Marjorie completed each question of this quiz, she placed that question's correct statement or best answer's letter in parentheses.

1. In developing his periodic table, Mendeleev's observations were
 - a. periodic elements.
 - b. elements in the liquid state.
 - c. isotopes.
 - d. **properties.**
2. Mendeleev's periodic table showed that all elements in each of the
 - a. groups shared many features, but grouped together elements with
 similar
 - b. **properties.**
 - c. similar elements.
 - d. isotopes.
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. ions.
 - b. unknown elements.
 - c. unstable isotopes.
 - d. **undiscovered elements.**
4. The periodic table came before a periodic table based on increasing
 - a. atomic number was
 - b. **discovered.**
 - c. Mendeleev.
 - d. classification.
5. The periodic law describes the physical and chemical properties of
 - a. **elements.**
 - b. isotopes.
 - c. ions.
 - d. compounds.

Periodic Law Section Quiz Answers

Shasha Hu

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the name bar.

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