

Percent Yield

1. In a certain chemical reaction, 100 g of reactant A reacts with 100 g of reactant B to produce 100 g of product C. What is the percent yield of product C? $\text{Percent Yield} = \frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100\%$
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John A. Neeley, Edward B. Giddings

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