

STUDY GUIDE 11: Oxidation and Reduction (Chapter 12)

Remember: Oxidation = Loss of Electrons

Oxidation: loss of electrons; decrease in O.N. number (including 0, 0, 0 bonds)

Reduction: gain of electrons; increase in O.N. number; decrease in O.N. bonds



Be careful: some compounds contain both oxidized and reduced elements, but the single element is

oxidized



Remember:

REDOX REACTION

Redox reactions involve the

Remember:

Remember: In a redox reaction, the oxidation number of the element that is oxidized increases



Remember: In a redox reaction, the oxidation number of the element that is reduced decreases



Oxidation And Reduction Study Guide

Neil E. Schore



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