

# Periodic Trends Questions

5. An element with high ionization energy would most likely be?
- a. A nonmetal with low electronegativity
  - b. A nonmetal with high electronegativity
  - c. A metal with low electronegativity
  - d. A metal with high electronegativity
6. What happens to S when it becomes  $S^{2-}$ ?
- a. It loses two electrons and the radius increases
  - b. It loses two electrons and the radius decreases
  - c. It gains two electrons and the radius increases
  - d. It gains two electrons and the radius decreases

# Periodic Trends Extension Questions

**Wayne C. Thompson**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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