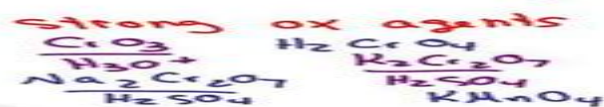


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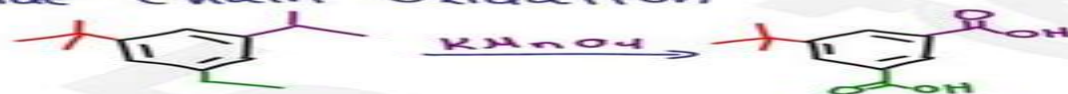
Oxidation = Gain O bonds, Lose H bonds



Alkene Oxidation (also alkynes)



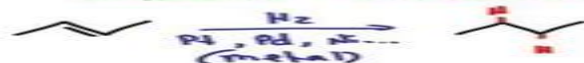
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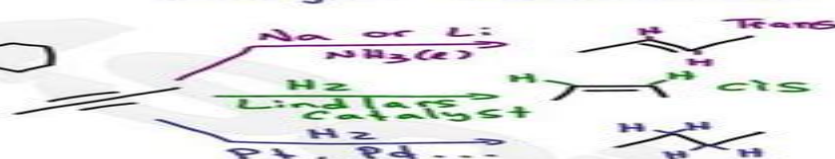
only 1° & 2°
 get oxidized
 3° = N/A

Reduction = Gain H bonds, Lose O bonds

Alkene Reduction



Alkyne Reduction



Birch Reduction



Reduction With Metal Hydrides



LiAlH₄ = strong



NaBH₄ = weak



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