Naming Compounds Containing Polyatomic Ions

Use the following list of polyatomic ions to name the compounds given.

	OH NO ₂ C ₂ H ₂ O ₂	hydroxide nitrate acetate	CIO ₃ HCO ₃ SO ₄ CO ₃	chlorate bicarbonate sulfate carbonate	SO, PO, NH,	sulfite phosphate ammonium
1.	CuSO			26. NaOH		
22.	KCIO,			27. NaNO,		
3.	NH4OH					
4.	K2CO3					
5.	Na ₃ SO,					
6.	KC2H2O2					
7.	NH,Cl					
8.	Mg(OH)2					
9.	$Zn(NO_a)_2$		v. 189 5			
10.	MgSO					
II.	CaCO ₂					
12.	кон					
13.	Ca(OH) ₂					
1.4.	K ₂ CO ₃					
15.	AIPO					
16.	Na ₃ PO ₄					
17.	MgSO,					
18.	AgNO ₃					
19.	Na ₂ CO ₅			44. NiOH		
				45. ZnSO.		
21.	Ca(NO ₃) ₂					
22.	BaSO,					
24.	LiNO ₃					

Polyatomic Ions Naming Answer

Michael Mosher, Paul Kelter

Polyatomic Ions Naming Answer:

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examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students Introduction to Chemistry Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools. The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry **Chemistry** Greg Curran, 2025-09-12 Homework Helpers Chemistry is a user friendly review book that will make every student or parent trying to help their child feel like he or she has a private Chemistry tutor Concepts are explained in clear easy to understand language and problems are worked out with step by step methods that are easy to follow Each lesson comes with numerous review questions and answer keynotes that explain each correct answer and why it s correct This book covers all of the topics in a typical one year Chemistry curriculum including A systematic approach to problem solving conversions and the use of units Naming compounds writing formulas and balancing chemical equations Gas laws chemical kinetics acids and bases electrochemistry and more While Homework Helpers Chemistryis an excellent review for any standardized Chemistry test including the SAT II its real value is in providing support and guidance during the year s entire course of study eBook: General, Organic and Biological Chemistry 2e SMITH, 2012-02-16 eBook General Organic and Biological Chemistry 2e Mapping College Chemistry Stephen DeMeo, 2019-03-01 This text is a chemistry problem solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first year college General Chemistry course The book incorporates a chemistry problem solving plan one that uses an innovative graphic organizer strategy The strategy successfully evaluated with students combines problem solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry Topical problem solving will focus on limiting reactant stoichiometry identifying types of chemical reactions

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