

Chapter 9 Physical Science Study Guide:

- Know all vocab (on Quizlet)
- Know/describe the different parts of an atom
- Describe the atomic model timeline
 - Scientists, theories, discoveries, models, experiments
- Know what quarks are
- Know what happens to a neutral atom when the number of protons, neutrons, or electrons change
- Be able to describe the three types of nuclear decay

Different Parts of an Atom:

- **Electrons** are very small particles inside an atom with one negative charge (-1). The closer electrons are to the nucleus, less energy.
- The **nucleus** is a small area in the center of the atom where most of its mass and positive charge is concentrated
- A **proton** is an atomic particle that has one positive charge (+1)
- A **neutron** is a neutral particle that exists in the nucleus of an atom
- An **electron cloud** is an area around an atomic nucleus where an electron is most likely to be located
- A **quark** is the smallest particle in an atom, and it is what protons and neutrons are made of

The Atomic Model Timeline:

DEMOCRITUS:

- He was a philosopher who challenged the popular idea of matter.
- His atomic theories were that atoms are small solid objects that cannot be divided, created, or destroyed. Also, that atoms are constantly moving in empty space, and that different types of matter are made of different types of atoms. His last theory is that the properties of the atoms determine the properties of matter.

DALTON:

- In the late 1700s, John Dalton revisited the idea of atoms.
- His atomic theories were that all matter is made of atoms that can't be divided, created, or destroyed. Also, that during a chemical reaction, atoms of one element can't be converted into atoms of another element. The third theory is that atoms of one element are identical to each other but different from atoms of another element. The last theory is that atoms combine in specific ratios.

Physical Science Chapter 9 Study Guide Answers

Julia Schneider



Physical Science Chapter 9 Study Guide Answers:

Barron's Science 360: A Complete Study Guide to Physics with Online Practice Barron's Educational Series, Kenneth Rideout, 2021-09-07 Barron's Math 360 Physics is your complete go to guide for everything physics This comprehensive guide is an essential resource for High school and college courses Homeschooling Virtual Learning Learning pods Inside you'll find Comprehensive Content Review Begin your study with the basic building blocks of physics and build as you go Topics include motion forces electricity magnetism and introduction to nuclear physics and much more Effective Organization Topic organization and simple lesson formats break down the subject matter into manageable learning modules that help guide a successful study plan customized to your needs Clear Examples and Illustrations Easy to follow explanations hundreds of helpful illustrations and numerous step by step examples make this book ideal for self study and rapid learning Practice Exercises Each chapter ends with practice exercises designed to reinforce and extend key skills and concepts These checkup exercises along with the answers and solutions will help you assess your understanding and monitor your progress Access to Online Practice Take your learning online for 50 practice questions designed to test your knowledge with automated scoring to show you how far you have come

Study Guide to Accompany: Fundamentals of Physical Science Six Edition Arthur Beiser, 1971

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Critical Appraisal of Physical Science as a Human Enterprise Mansoor Niaz,2009-02-07 It is generally believed that doing science means accumulating empirical data with no or little reference to the interpretation of the data based on the scientist s th retical framework or presuppositions Holton 1969a has deplored the widely accepted myth experimenticism according to which progress in science is presented as the inexorable result of the pursuit of logically sound conclusions from un biguous experimental data Surprisingly some of the leading scientists themselves Millikan is a good example have contributed to perpetuate the myth with respect to modern science being essentially empirical that is carefully tested experim tal facts free of a priori conceptions leading to inductive generalizations Based on the existing knowledge in a field of research a scientist formulates the guiding assumptions Laudan et al 1988 presuppositions Holton 1978 1998 and hard core Lakatos 1970 of the research program that constitutes the imperative of presuppositions which is not abandoned in the face of anomalous data Laudan and his group consider the following paraphrase of Kant by Lakatos as an important guideline philosophy of science without history of science is empty Starting in the 1960s this historical school

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