#### Unit 2: Atomic Scientists and Atomic Structure Unit II, LT # 1&2

#### Atomic Scientists and Atomic Structure Magic Squares

Directions: Put the number of the definition from the list below into the square with the appropriate term. Check your answers by adding the numbers to see if all the sums of all rows, both across and down add up to the same number, the Magic #.

Democritus	John Dalton	J.J. Thomson	Henry Moseley	Total
2		18	12	39
Ernest Rutherford 8	Proton 5	Atom	Henri Becquerel	39
Millikan 13	Neutron 17	Nucleus 6	Alpha Particle 3	_39
Electron	Model	Mendeleev 4	Electron Cloud	39
Total 39	39	39	39	

Magic Number: 39

- 1. Represented by a symbol; all are found on the Periodic Table
- 2. Made a mental model of the atom; Greek philosopher
- Used by Rutherford in his experiment; made of two protons and two neutrons (aka: He)
- 4. Considered to be the father of the periodic table.
- 5. The positive particle in the nucleus of an atom. (Not a fundamental particle)
- 6. The tiny positive core of an atom; contains protons and neutrons.
- Credited with the first atomic theory model of the atom which had four main postulates.
- 8. Discovered the nucleus using his gold foil experiment.
- 9. Current explanation of where electrons might be found in the atom.
- 10. Used by scientists to explain something we can not see or understand.
- 11. The smallest particle of an element that has the properties of that element.
- Arranged the periodic table in terms of the atomic number of the element.
- 13. Determined the exact charge of the electron with his oil drop experiment.
- Mass of protons and neutrons
- 15. Along with Marie and Pierre Curie, discovered radioactivity.
- 16. The negative particle that circles the nucleus

# **Periodic Table Magic Square Answer Key**

**Yijin Wang** 

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Natural and Unnatural Product Chemistry Bryan Hanley, 2025-01-26 This book examines and summarises the developments and changes in the approach of organic and natural product chemists over the past years presenting both the successes and inevitable missteps These illustrate the process from molecular isolation and structural identification synthesis and biosynthesis and future perspectives including the development of biological engineering The anticipated audience are chemistry literate scientists who are interested both in specific aspects of natural product chemistry and the illustrative process of trial error and success which defines the area **Match** ,2003 Chemical Heritage ,2008 **America's Story** An Introduction to Quantum Theory Frank S. Levin, 2002 An undergraduate introductory quantum mechanics textbook with a large number of figures and exercises NMR and the Periodic Table Robin Kingsley Harris, Brian E. Mann, 1978 Soviet Life ,1970-05 Scientific American ,1908 Introduction to Nuclear Reactor Physics Robert E. Masterson, 2017-11-22 INTRODUCTION TO NUCLEAR REACTOR PHYSICS is the most comprehensive modern and readable textbook for this course module It explains reactors fuel cycles radioisotopes radioactive materials design and operation Chain reaction and fission reactor concepts are presented plus advanced coverage including neutron diffusion theory The diffusion equation Fisk's Law and steady state time dependent reactor behavior Numerical and analytical solutions are also covered The text has full color illustrations throughout and a wide range of student learning features Horseless Carriage Days Hiram Percy Maxim, 1962

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