

# PERIODIC TABLE (ADVANCED)

Molecular Structure of Hydrogen



## Basic Information

Name: Hydrogen

Symbol: H

Atomic Number: 1

Atomic Mass: 1.00794 amu

Melting Point:  $-259.14^{\circ}\text{C}$  ( $-314.59995^{\circ}\text{K}$ ,  $-434.41203^{\circ}\text{F}$ )

Boiling Point:  $-252.87^{\circ}\text{C}$  ( $-253.05995^{\circ}\text{K}$ ,  $-423.166^{\circ}\text{F}$ )

Number of Protons/Electrons: 1

Number of Neutrons: 0

Classification: Non-metal

Crystal Structure: Hexagonal

Density @  $293^{\circ}\text{K}$ :  $8.9898\text{ g/cm}^3$

Color: colorless

## Atomic Structure

Number of Energy Levels: 1

First Energy Level: 1

## Isotopes

| Isotope | Half Life  |
|---------|------------|
| H-1     | Stable     |
| H-2     | Stable     |
| H-3     | 12.5 years |

## Facts

Name Origin: Greek words *hydro* (water) and *genes* (generated)

Uses: Balloons, metal refining

Obtained From: natural, all gas water

Molecular Structure of Carbon



## Name: Carbon

Symbol: C

Atomic Number: 6

Atomic Mass: 12.0087 amu

Melting Point:  $3500^{\circ}\text{C}$  ( $3773.15^{\circ}\text{K}$ ,  $6312.8^{\circ}\text{F}$ )

Boiling Point:  $4602.1^{\circ}\text{C}$  ( $5100.15^{\circ}\text{K}$ ,  $8320.6^{\circ}\text{F}$ )

Number of Protons/Electrons: 6

Number of Neutrons: 6

Classification: Non-metal

Crystal Structure: Hexagonal

Density @  $293^{\circ}\text{K}$ :  $2.62\text{ g/cm}^3$

Color: May be black

## Atomic Structure

Number of Energy Levels: 2

First Energy Level: 2

Second Energy Level: 4

## Isotopes

| Isotope | Half Life    |
|---------|--------------|
| C-11    | 20.3 minutes |
| C-12    | Stable       |
| C-13    | Stable       |
| C-14    | 5730 years   |
| C-15    | 2.3 seconds  |

## Facts

Uses: steel, flame

Obtained From: burning with insufficient oxygen

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**R Sandford**



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