

Practice Final Exam, Chemistry 2220, Organic Chem II

1. Rank the designated protons by chemical shift (δ), highest δ first.



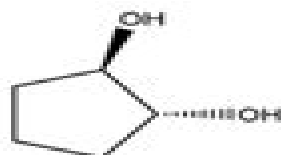
- A. $\text{IV} > \text{III} > \text{II} > \text{I}$ B. $\text{IV} > \text{II} > \text{III} > \text{I}$ C. $\text{III} > \text{IV} > \text{II} > \text{I}$ D. $\text{III} > \text{IV} > \text{I} > \text{II}$

2. What are the products, if any, expected from the following reaction?



- $\text{CH}_3\text{OCH}_3 + \text{NaHS}$ $\text{CH}_3\text{SCH}_3 + \text{NaOH}$ $\text{CH}_3\text{O}-\text{H} + \text{CH}_3\text{S}^-\text{Na}^+$ no reaction
 A. B. C. D.

3. What is the IUPAC name for the following compound?



- A. cis-1,2-cyclopentanediol
 B. meso-1,2-cyclopentanediol
 C. (1R,2R)-1,2-cyclopentanediol
 D. (1R,2S)-1,2-cyclopentanediol

4. Which of the following explains why cyclohexanol has a pK_a of 18 and phenol has a pK_a of 10?

- A. Phenolate is a stronger base than the conjugate base of cyclohexanol
 B. The conjugate base of cyclohexanol is resonance stabilized
 C. The conjugate base of phenol is resonance stabilized
 D. Phenol is a weaker acid than cyclohexanol

Odysseyware Final Exam For Chemistry

Hussin A.Rothana



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