

(iii) What does $x\text{H}_2\text{O}$ represent in the formula $\text{Al}_2(\text{SO}_4)_3 \cdot x\text{H}_2\text{O}$?

Water of crystallisation.

[1]

H_2O crystals to constant mass.

has a mass of 5.76 g.

value of x .

+



$$\frac{5.76\text{g}}{18}$$

$$= 0.32\text{mol}$$

$$\frac{0.02}{=}$$

$$16$$

$x =$



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Dave Gent

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'Dave Gent' text bar.

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Table of Contents Ocr June 2013 F321

1. Understanding the eBook Ocr June 2013 F321
 - The Rise of Digital Reading Ocr June 2013 F321
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr June 2013 F321
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr June 2013 F321
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr June 2013 F321
 - Personalized Recommendations
 - Ocr June 2013 F321 User Reviews and Ratings
 - Ocr June 2013 F321 and Bestseller Lists
5. Accessing Ocr June 2013 F321 Free and Paid eBooks
 - Ocr June 2013 F321 Public Domain eBooks
 - Ocr June 2013 F321 eBook Subscription Services
 - Ocr June 2013 F321 Budget-Friendly Options
6. Navigating Ocr June 2013 F321 eBook Formats
 - ePub, PDF, MOBI, and More
 - Ocr June 2013 F321 Compatibility with Devices
 - Ocr June 2013 F321 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr June 2013 F321
 - Highlighting and Note-Taking Ocr June 2013 F321
 - Interactive Elements Ocr June 2013 F321
8. Staying Engaged with Ocr June 2013 F321

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Ocr June 2013 F321
- 9. Balancing eBooks and Physical Books Ocr June 2013 F321
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr June 2013 F321
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ocr June 2013 F321
 - Setting Reading Goals Ocr June 2013 F321
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ocr June 2013 F321
 - Fact-Checking eBook Content of Ocr June 2013 F321
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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