

Recrystallization Pre-Lab

Introduction:

The purpose of this experiment is to recrystallize phenanthrene and benzoic acid to purify the compounds. A compound is dissolved in hot solvent in recrystallization which in this experiment, phenanthrene is dissolved in a solvent pair of water and ethanol, and benzoic acid is dissolved in hexane. The solution will be cooled once completely mixed with these solvents. By doing this, the solute's solubility will decrease and will lead to the formation of benzoic acid and phenanthrene crystal.

Physical Data:

Name	MW (g/mol)	MP (°C)	BP (°C)	Density(g/ml)	Solubility	Hazards
Benzoic acid	122.12	122	250	1.2659	Soluble in hexane, acetone; less soluble in water	Cause skin irritation harmful if consumed
Phenanthrene	178.23	101	332	1.18	Insoluble in water, soluble in benzene, chloroform, acetic acid	Moderate health problems harmful if consumed
Ethanol	46.07	-114.1	79	0.8	Soluble in water and benzene	Highly flammable and toxic
Hexane	86.18	-95	69	0.7	Insoluble in water by soluble in ethanol	Highly flammable

Recrystallization Lab Report Organic Chemistry

Tobias Bleicker



Recrystallization Lab Report Organic Chemistry:

Unveiling the Magic of Words: A Review of "**Recrystallization Lab Report Organic Chemistry**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Recrystallization Lab Report Organic Chemistry**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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