

# Memorandum Grade 11 p/sciences

Physical Sciences Gr11/P2

2  
NSC

September

## QUESTION 1

1.1 A ✓✓

(2)

1.2 B ✓✓

(2)

1.3 C & ✓✓

(2)

1.4 B & ✓✓

(2)

1.5 C & ✓✓

(2)

1.6 D ✓✓

(2) ~~(10)~~ (12)

## QUESTION 2

2.1 The mass of one mole of a substance measured in  $\text{g}\cdot\text{mol}^{-1}$  ✓✓ (2 or 0)

(2)

2.2

| Marking criteria                                                                                             |                                                                           |                           |
|--------------------------------------------------------------------------------------------------------------|---------------------------------------------------------------------------|---------------------------|
| (a) Any formula $m = nM$ or $N = nN_A$ ✓                                                                     |                                                                           |                           |
| (b) Substitution of $3,114 \times 10^{24}$ and $6,02 \times 10^{23}$ into $N = nN_A$ ✓                       |                                                                           |                           |
| (c) Substitution of $1 \text{ g}\cdot\text{mol}^{-1}$ into $m = nM$ ✓                                        |                                                                           |                           |
| (d) Mole ratio 3:6:1 ✓                                                                                       |                                                                           |                           |
| (e) $X = 3$ ✓, $Y = 6$ ✓, $Z = 1$ ✓                                                                          |                                                                           |                           |
| C                                                                                                            | H                                                                         | O                         |
| $n = \frac{N}{N_A}$ ✓<br>$n = \frac{3,114 \times 10^{24}}{6,02 \times 10^{23}}$ ✓<br>$n = 5,173 \text{ mol}$ | $n = \frac{m}{M}$<br>$n = \frac{10,345}{1}$ ✓<br>$n = 10,345 \text{ mol}$ | $n = 1,724 \text{ mol}$   |
| $\frac{5,173}{1,724} = 3$                                                                                    | $\frac{10,345}{1,724} = 6$                                                | $\frac{1,724}{1,724} = 1$ |
| 3:6:1 ✓                                                                                                      |                                                                           |                           |
| $X = 3$ ✓                                                                                                    | $Y = 6$ ✓                                                                 | $O = 1$ ✓                 |

Accept values indicated in the empirical formula:  $\text{C}_3\text{H}_6\text{O}$  ✓✓✓

(7)

# Physical Science Memoradam Grade 11 September 2013

**K Morrison**



**Physical Science Memoradam Grade 11 September 2013:**

## Decoding **Physical Science Memoradam Grade 11 September 2013**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Physical Science Memoradam Grade 11 September 2013**," a mesmerizing literary creation penned by a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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### **Table of Contents Physical Science Memoradam Grade 11 September 2013**

1. Understanding the eBook Physical Science Memoradam Grade 11 September 2013
  - The Rise of Digital Reading Physical Science Memoradam Grade 11 September 2013
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Memoradam Grade 11 September 2013
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in a Physical Science Memoradam Grade 11 September 2013
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Memoradam Grade 11 September 2013
  - Personalized Recommendations

- Physical Science Memoradam Grade 11 September 2013 User Reviews and Ratings
- Physical Science Memoradam Grade 11 September 2013 and Bestseller Lists
- 5. Accessing Physical Science Memoradam Grade 11 September 2013 Free and Paid eBooks
  - Physical Science Memoradam Grade 11 September 2013 Public Domain eBooks
  - Physical Science Memoradam Grade 11 September 2013 eBook Subscription Services
  - Physical Science Memoradam Grade 11 September 2013 Budget-Friendly Options
- 6. Navigating Physical Science Memoradam Grade 11 September 2013 eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science Memoradam Grade 11 September 2013 Compatibility with Devices
  - Physical Science Memoradam Grade 11 September 2013 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Memoradam Grade 11 September 2013
  - Highlighting and Note-Taking Physical Science Memoradam Grade 11 September 2013
  - Interactive Elements Physical Science Memoradam Grade 11 September 2013
- 8. Staying Engaged with Physical Science Memoradam Grade 11 September 2013
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Memoradam Grade 11 September 2013
- 9. Balancing eBooks and Physical Books Physical Science Memoradam Grade 11 September 2013
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Memoradam Grade 11 September 2013
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Memoradam Grade 11 September 2013
  - Setting Reading Goals Physical Science Memoradam Grade 11 September 2013
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Memoradam Grade 11 September 2013
  - Fact-Checking eBook Content of Physical Science Memoradam Grade 11 September 2013

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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