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Exceed to Succeed

Worksheet: Periodic Trends

1. ATOMIC RADIUS

For each of the following sets of atoms, rank the atoms from smallest to largest atomic radius.

- a. Li, C, F
- b. Li, Na, K
- c. Ge, P, O
- d. C, N, Al
- e. Al, Cl, Ga

2. IONIC RADIUS

For each of the following sets of ions, rank them from smallest to largest ionic radius.

- a. Mg^{2+} , Si^{4+} , S^{2-}
- b. Mg^{2+} , Ca^{2+} , Ba^{2+}
- c. F^- , Cl^- , Br^-
- d. Ba^{2+} , Cu^{2+} , Zn^{2+}
- e. Si^{4+} , P^{3-} , O^{2-}

3. IONIZATION ENERGY

For each of the following sets of atoms, rank them from lowest to highest ionization energy.

- a. Mg, Si, S
- b. Mg, Ca, Ba
- c. F, Cl, Br
- d. Ba, Cu, Ne
- e. Si, P, He

4. ELECTRONEGATIVITY

For each of the following sets of atoms, rank them from lowest to highest electronegativity.

- a. Li, C, N
- b. C, O, Ne
- c. Si, P, O
- d. K, Mg, P
- e. S, F, He

Periodic Trends Worksheet Using The Periodic Law Answers

Alastair M. Morrison



Periodic Trends Worksheet Using The Periodic Law Answers:

Periodic Trends Worksheet Using The Periodic Law Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Periodic Trends Worksheet Using The Periodic Law Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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