

Extension Questions

15. During this activity you may have noticed that not all of the data provided in the models followed the trends.
- Identify two places in Model 1 where the property listed does not fit the trend identified in this activity.
 - Why is it still beneficial for chemists to understand as many periodic trends as they can?
 - Propose an explanation for one of the exceptions you identified in part *a*. Use your knowledge of atomic structure and Coulombic attraction in your hypothesis.
16. Rank the following elements from smallest to largest electronegativity based on the trends you have discovered thus far in the periodic table: barium (atomic number 56), bromine (atomic number 35), and iron (atomic number 26). Explain your reasoning.

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RJ Shavelson



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