

## Polyatomic Ions Worksheet Answer Key

### Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

sulfate	<u><math>\text{SO}_4^{2-}</math></u>	<u><math>\text{CO}_3^{2-}</math></u>	carbonate
nitrite	<u><math>\text{NO}_2^-</math></u>	<u><math>\text{MnO}_4^{1-}</math></u>	manganate
perphosphate	<u><math>\text{PO}_4^{3-}</math></u>	<u><math>\text{SO}_5^{2-}</math></u>	persulfate
hypoiodite	<u><math>\text{IO}^-</math></u>	<u><math>\text{BrO}_2^{1-}</math></u>	bromite
chlorite	<u><math>\text{ClO}_2^-</math></u>	<u><math>\text{CO}_4^{2-}</math></u>	percarbonate
phosphite	<u><math>\text{PO}_3^{3-}</math></u>	<u><math>\text{PO}_5^{3-}</math></u>	perphosphate
percarbonate	<u><math>\text{CO}_4^{2-}</math></u>	<u><math>\text{ClO}^{1-}</math></u>	hypochlorite
bromate	<u><math>\text{BrO}_3^-</math></u>	<u><math>\text{IO}_2^{1-}</math></u>	iodite
hyposulfite	<u><math>\text{SO}_2^{2-}</math></u>	<u><math>\text{PO}_4^{3-}</math></u>	phosphate
permanganate	<u><math>\text{MnO}_4^-</math></u>	<u><math>\text{NO}_2^{1-}</math></u>	nitrite
carbonite	<u><math>\text{CO}_2^{2-}</math></u>	<u><math>\text{SO}_4^{2-}</math></u>	sulfate

2. Name or write the formula for the following Type I polyatomic ionic compounds

beryllium hydroxide	<u><math>\text{Be}(\text{OH})_2</math></u>	<u><math>\text{Ba}(\text{IO}_3)_2</math></u>	barium iodate
sodium nitrite	<u><math>\text{NaNO}_2</math></u>	<u><math>\text{Ga}(\text{CNO})_3</math></u>	gallium cyanate
ammonium chloride	<u><math>\text{NH}_4\text{Cl}</math></u>	<u><math>\text{Ag}_2\text{SO}_3</math></u>	silver sulfite
calcium bisulfate	<u><math>\text{Ca}(\text{HSO}_4)_2</math></u>	<u><math>\text{MgCO}</math></u>	magnesium hypocarbonite
rubidium perchlorate	<u><math>\text{RbClO}_4</math></u>	<u><math>\text{NH}_3\text{NO}_2</math></u>	ammonium nitrite
strontium sulfite	<u><math>\text{SrSO}_3</math></u>	<u><math>\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3</math></u>	aluminum acetate
aluminum acetate	<u><math>\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3</math></u>	<u><math>\text{SrSO}_5</math></u>	strontium persulfate
ammonium nitrate	<u><math>\text{NH}_4\text{NO}_3</math></u>	<u><math>\text{RbClO}_2</math></u>	rubidium chlorite
magnesium hypocarbonite	<u><math>\text{MgCO}</math></u>	<u><math>\text{Ca}(\text{HSO}_4)_2</math></u>	calcium bisulfate
silver hyposulfite	<u><math>\text{Ag}_2\text{SO}_2</math></u>	<u><math>\text{NH}_4\text{Cl}</math></u>	ammonium chloride

# Polyatomic Ions Worksheet Answer Key Unit 3

**Minjie Lin**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

### **Polyatomic Ions Worksheet Answer Key Unit 3:**

## Unveiling the Magic of Words: A Review of "**Polyatomic Ions Worksheet Answer Key Unit 3**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Polyatomic Ions Worksheet Answer Key Unit 3**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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