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Exceed to Succeed

Worksheet: Periodic Trends

1. ATOMIC RADIUS

For each of the following sets of atoms, rank the atoms from smallest to largest atomic radius.

- a. Li, C, F
- b. Li. Na. K
- c. Ge. P. O.
- d. C. N. Al
- e. Al. Cl. Ga.

IONIC RADIUS 2...

For each of the following sets of ions, rank them from smallest to largest ionic radius.

- a. Mg²⁺, Si⁴⁻, S²⁻ b. Mg²⁺, Ca²⁺, Ba²⁺
- c. F. Cl. Br
- d. Ba²⁺, Cu²⁺, Zn²⁺ e. Si⁴⁻, P³⁻, O²⁻

IONIZATION ENERGY 3...

For each of the following sets of atoms, rank them from lowest to highest ionization energy.

- a. Mg, Si, S
- b. Mg, Ca, Ba
- c. F. Cl. Br
- d. Ba. Cu. Ne
- e. Si. P. He

46. ELECTRONEGATIVITY

For each of the following sets of atoms, rank them from lowest to highest electronegativity.

- a. Li, C, N
- b. C. O. Ne
- c. Si. P. O
- d. K. Mg, P
- e. S. F. He.

Periodic Trends Review Worksheet Answer Sheet

Camilla Rothe

Periodic Trends Review Worksheet Answer Sheet:

Periodic Trends Review Worksheet Answer Sheet Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Periodic Trends Review Worksheet Answer Sheet**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

https://crm.avenza.com/public/publication/index.jsp/nikon%2070%20300mm%20manual.pdf

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