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Periodic Trends and Compounds Review Worksheet

Case Report

Use any material as your reference as you might deem necessary to answer the questions below in exception of asking someone else to give you the answer.

1. Which group tends to form 1+ ions? _____
2. Which group tends to form 2+ ions? _____
3. Which group tends to form 3+ ions? _____
4. Which group tends not to form ions or at all? _____

5. Locate the alkali metals, alkaline earth metals, halogens, noble gases, actinides, lanthanides, main group elements, semimetals (metalloids), transition metal s and inner transition metals on the periodic table below.

Detailed Timeline of Task Completion

Task	Start Time	End Time	Duration (h)
Task A	0	10	10
Task B	0	15	15
Task C	0	20	20
Task D	10	22	12
Task E	15	33	18
Task F	20	45	25
Task G	22	32	10
Task H	33	48	15
Task I	45	65	20
Task J	32	44	12
Task K	48	66	18
Task L	65	90	25
Task M	44	54	10
Task N	66	81	15
Task O	90	110	20
Task P	54	66	12
Task Q	81	99	18
Task R	110	135	25
Task S	66	76	10
Task T	99	114	15
Task U	135	155	20
Task V	76	88	12
Task W	114	132	18
Task X	155	180	25
Task Y	88	98	10
Task Z	132	147	15

6. What repeating pattern is noticeable regarding the ionization energies of different elements? How does the periodic table account for this pattern?

7. Explain the patterns for acid ionizability and solubility seen. Explain the observed solubility

Periodic Trends And Compounds Review

Jicheng Xie



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