

PERIODIC TABLE (BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																
1 H Hydrogen 1.008	2 He Helium 4.003											10 Ne Neon 20.180	11 Na Sodium 22.990	12 Mg Magnesium 24.305	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.065	17 Cl Chlorine 35.453	18 Ar Argon 39.948													
3 Li Lithium 6.941	4 Be Beryllium 9.012											19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 52.004	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.630	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80				
5 Rb Rubidium 85.468	6 Sr Strontium 87.62	37 Y Yttrium 88.906	38 Zr Zirconium 91.224	39 Nb Niobium 92.906	40 Mo Molybdenum 95.94	41 Tc Technetium (98)	42 Ru Ruthenium 101.07	43 Rh Rhodium 102.91	44 Pd Palladium 106.42	45 Ag Silver 107.87	46 Cd Cadmium 112.41	47 In Indium 114.82	48 Sn Tin 118.71	49 Sb Antimony 121.76	50 Te Tellurium 127.6	51 I Iodine 126.90	52 Xe Xenon 131.29	53 Ba Barium 137.33	54 La Lanthanum 138.91	55 Ce Cerium 140.12	56 Pr Praseodymium 140.91	57 Nd Neodymium 144.24	58 Pm Promethium (145)	59 Sm Samarium 150.36	60 Eu Europium 151.96	61 Gd Gadolinium 157.25	62 Tb Terbium 158.93	63 Dy Dysprosium 162.50	64 Ho Holmium 164.93	65 Er Erbium 167.26	66 Tm Thulium 168.93	67 Yb Ytterbium 173.05	68 Lu Lutetium 174.96
7 Fr Francium (223)	8 Ra Radium (226)	87 Fr Francium (223)	88 Ra Radium (226)	89 Ac Actinium (227)	90 Th Thorium (232)	91 Pa Protactinium (231)	92 U Uranium (238)	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (260)	104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (263)	107 Bh Bohrium (264)	108 Hs Hassium (265)	109 Mt Meitnerium (266)	110 Ds Darmstadtium (271)	111 Rg Roentgenium (272)	112 Cn Copernicium (285)	113 Nh Nihonium (284)	114 Fl Flerovium (289)	115 Mc Moscovium (288)	116 Lv Livermorium (293)	117 Ts Tennessine (294)	118 Og Oganesson (294)

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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57 La Lanthanum 138.91	58 Ce Cerium 140.12	59 Pr Praseodymium 140.91	60 Nd Neodymium 144.24	61 Pm Promethium (145)	62 Sm Samarium 150.36	63 Eu Europium 151.96	64 Gd Gadolinium 157.25	65 Tb Terbium 158.93	66 Dy Dysprosium 162.50	67 Ho Holmium 164.93	68 Er Erbium 167.26	69 Tm Thulium 168.93	70 Yb Ytterbium 173.05	71 Lu Lutetium 174.96		
87 Fr Francium (223)	88 Ra Radium (226)	89 Ac Actinium (227)	90 Th Thorium (232)	91 Pa Protactinium (231)	92 U Uranium (238)	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (260)

Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic forms (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος, lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

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1. Understanding the eBook Periodic Table Basic Speedy Study Guides Speedy Publishing
 - The Rise of Digital Reading Periodic Table Basic Speedy Study Guides Speedy Publishing
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Basic Speedy Study Guides Speedy Publishing
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Platform
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Basic Speedy Study Guides Speedy Publishing
 - Personalized Recommendations
 - Periodic Table Basic Speedy Study Guides Speedy Publishing User Reviews and Ratings
 - Periodic Table Basic Speedy Study Guides Speedy Publishing and Bestseller Lists
5. Accessing Periodic Table Basic Speedy Study Guides Speedy Publishing Free and Paid eBooks
 - Periodic Table Basic Speedy Study Guides Speedy Publishing Public Domain eBooks
 - Periodic Table Basic Speedy Study Guides Speedy Publishing eBook Subscription Services
 - Periodic Table Basic Speedy Study Guides Speedy Publishing Budget-Friendly Options
6. Navigating Periodic Table Basic Speedy Study Guides Speedy Publishing eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Basic Speedy Study Guides Speedy Publishing Compatibility with Devices
 - Periodic Table Basic Speedy Study Guides Speedy Publishing Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Basic Speedy Study Guides Speedy Publishing
 - Highlighting and Note-Taking Periodic Table Basic Speedy Study Guides Speedy Publishing
 - Interactive Elements Periodic Table Basic Speedy Study Guides Speedy Publishing
8. Staying Engaged with Periodic Table Basic Speedy Study Guides Speedy Publishing

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Basic Speedy Study Guides Speedy Publishing
9. Balancing eBooks and Physical Books Periodic Table Basic Speedy Study Guides Speedy Publishing
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Basic Speedy Study Guides Speedy Publishing
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Basic Speedy Study Guides Speedy Publishing
- Setting Reading Goals Periodic Table Basic Speedy Study Guides Speedy Publishing
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Basic Speedy Study Guides Speedy Publishing
- Fact-Checking eBook Content of Periodic Table Basic Speedy Study Guides Speedy Publishing
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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