energy:	he element in each of the t	
a) lithium	or boron	c) cesium or lead
b) magnes	ium or strontium	d) sodium or argon
1. Arrange the	e following elements in ord	ler of increasing ionization energy:
a) Sr, Be, A	Mg	
b) Bi, Cs,	Ba	
c) Na, Al,	s	
2. Why is ther elements in		ween the 1st and 2nd ionization energies for the
Name:		Date: Class period:
	"lonic Size ar	Periodic Trends nd E.N. Trend" - WS #4
Pirections: Pla	ease answer each fill in the	e blank with the best answer. O
. A cation is	an ion with an overall	charge.
	The second secon	
. Cations are	made when an atom	<u>@</u>
		than the size of the neutral
. The size of		
. The size of atom of the	a positive ion is always e same element.	
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