



Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = 30.0 \text{ g/mol}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H₂O) gmm:

$$(2)(1.0) + (1)(16.0) = 18.0 \text{ g/mol}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = 88.9\%$$

$$\frac{45 \text{ g NO} \times \frac{1 \text{ mol NO}}{30 \text{ g NO}} \times \frac{4 \text{ mol NO}}{4 \text{ mol NO}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times 18 \text{ g H}_2\text{O}}{1} = 6.75 \text{ g H}_2\text{O}$$

Percent Yield Problems Answers

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Percent Yield Problems Answers:

Whispering the Secrets of Language: An Psychological Journey through **Percent Yield Problems Answers**

In a digitally-driven earth where monitors reign great and instant communication drowns out the subtleties of language, the profound secrets and psychological nuances concealed within words often move unheard. However, situated within the pages of **Percent Yield Problems Answers** a charming literary treasure sporting with raw thoughts, lies an extraordinary journey waiting to be undertaken. Penned by an experienced wordsmith, this charming opus encourages visitors on an introspective journey, lightly unraveling the veiled truths and profound affect resonating within the very material of each and every word. Within the emotional depths of this emotional evaluation, we will embark upon a genuine exploration of the book is key themes, dissect its interesting writing type, and succumb to the powerful resonance it evokes serious within the recesses of readers hearts.

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