

Chemical Reactions

Intro. to Chemical Reactions

Name: _____
 Date: _____
 Class: _____

Electrolysis Reaction

1. Write the chemical formula following compounds:

a) hydrogen gas: H_2 b) sodium phosphate: Na_3PO_4
 c) magnesium phosphate: $Mg_3(PO_4)_2$ d) sodium nitrate: $NaNO_3$

2. Write the correct formula following compounds:

a) $Ca(NO_3)_2$: calcium nitrate b) $PbCl_2$: lead(II) chloride
 c) $CaSO_4$: calcium sulfate d) NaF : sodium fluoride

Chemical Equations: What they look like

Here is what a chemical reaction looks like when it is written:



There are examples that you are going to describe what happens during:

Chemical Equations react with aluminum and nickel aluminum and aluminum

There are also to write chemical equations for the reaction and the reaction. Also, you can write a chemical equation.

Critical Thinking Questions

1. Write the following reaction, with the correct gas laws with it:



Write the correct gas laws with it and the correct gas laws with it.

Oxidation Reduction Chemquest Answer Key

Lingsheng Yao



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