

Directions: Choose the letter that corresponds to the correct answer.

1. Which of the following can be considered as Dobereiner's triad?
 - a. He: 4.0, Li: 6.9, Na: 23.0
 - b. N: 14.0, P: 40.0, At: 217.0
 - c. Ne: 20.2, Ar: 39.9, Kr: 83.8
 - d. Ca: 40.1, Sr: 87.6, Ba: 137.3
2. Which of the following elements does not belong to the group?
 - a. N
 - b. S
 - c. Po
 - d. Te
3. Which of the following is/are TRUE about boron and oxygen?
 - a. both are metalloids
 - b. boron is smaller than oxygen
 - c. boron is more electronegative than oxygen
 - d. higher energy is required to eject an electron from the valence shell of boron
 - e. all of the above
 - f. none of the above

For 4 and 5, consider Na, Mg, Al, and Si.

4. The element that ejects one electron from its valence shell most readily is _____.
 - a. Na
 - b. Mg
 - c. Al
 - d. Si
5. Which element has the most negative electron affinity?
 - e. Na
 - f. Mg

Periodic Trends Practice Questions

Thomas Bond, Chris Hughes



Periodic Trends Practice Questions:

Decoding **Periodic Trends Practice Questions**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Periodic Trends Practice Questions**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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