

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The diagram shows a crossword puzzle grid with the following letters placed in specific cells:

- Across:** A (1,1), B (1,10), C (2,1), D (2,9), E (2,10), F (3,2), G (3,7), H (3,9), I (3,10), J (4,1), K (4,10), L (4,2), M (4,7), N (4,9), O (4,10), P (4,11), Q (5,1), R (5,2), S (5,7), T (5,8), U (5,9), V (6,1), W (6,9), X (6,1), Y (6,9), Z (6,11).
- Down:** A (1,1), B (1,10), C (2,1), D (2,9), E (2,10), F (3,2), G (3,7), H (3,9), I (3,10), J (4,1), K (4,10), L (4,2), M (4,7), N (4,9), O (4,10), P (4,11), Q (5,1), R (5,2), S (5,7), T (5,8), U (5,9), V (6,1), W (6,9), X (6,1), Y (6,9), Z (6,11).

Arrows point from the letters R and Z to a separate section below the grid, which contains the letters V, W, X, and Y.

List elements in *alphabetical order* when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- _____ 4) Which letter labels the halogen element with the lowest atomic number?
- _____ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- _____ 6) Which of these elements is/are NOT transition elements/metals. (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

Periodic Tables Answer Key

A Gutmann



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