## **Periodic Table Extra Practice Worksheet**

Directions: Answer the following questions using the Periodic Table below.

IA.							VIII
1 H 1	IIA	IIIB	IVB	VB	VIB	VIIB	2 He 4
3 Li 7	Be 9	5 B 11	6 C 12	7 N 14	8 O 16	9 F 19	10 Ne 20
11 Na 23	12 Mg 24	13 Al 27	14 Si 28	15 P 31	16 S 32	17 CI 35	18 Ar 40

1. How many protons, neutrons, and electrons are in each of the following atoms?

Element	Boron (B)	Sodium (Na)	Phosphorus (P)
Protons	5	11.	15
Neutrons	6	12	16
Electrons	5	11	15

2. Which diagram best shows nucleus of beryllium? Circle the correct answer.

$$\begin{pmatrix}
n = 4 \\
e = 5
\end{pmatrix}$$
 $\begin{pmatrix}
p = 4 \\
e = 4
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 5 \\
n = 4
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 5 \\
n = 5
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 4 \\
n = 5
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 9 \\
n = 4
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 4 \\
n = 5
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 4 \\
n = 4
\end{pmatrix}$ 
 $\begin{pmatrix}
p = 9 \\
n = 4
\end{pmatrix}$ 

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3. Fill in the chart below from the data provided.

ELEMENT	ATOMIC NUMBER	MASS NUMBER	NUMBER OF PROTONS	NUMBER OF NEUTRONS	NUMBER OF ELECTRONS
Aluminum	a.13	27	b.13	c.14	d.13
Hydrogen	e.1	f.1	1	g.0	h.1
Argon	i.18	j.40	k.18	1.22	18
Magnesium	12	m.24	n.12	0.12	p.12

# **Periodic Table Practice Answers**

John T. Moore, Chris Hren, Peter J. Mikulecky

#### **Periodic Table Practice Answers:**

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