

Periodic Trends Worksheet

Using a few minutes before the beginning activity, complete the following questions. Please post your answers on the board when you are done.

Additional help on the periodic trends can be found on page 104 of the student textbook. <https://www.ck12.org/c/chemistry/periodic-table-of-elements-2/learn/lesson/Periodic-Trends-2/>

1. What does electronegativity measure and how does it change across the periodic table? (electronegativity is largest) in which direction, left-to-right, up-to-down, and bottom-to-top?
2. Electronegativity increases from left to right across the periodic table.
3. Electronegativity increases from left to right across the periodic table in which group:
 - a. Na or Li
 - b. Cl or F
 - c. Cl or Br
4. Electronegativity increases from left to right across the periodic table in which group:
 - a. R or Ca
 - b. Ca or Sr
 - c. Li or Rb
5. Electronegativity increases from left to right across the periodic table in which group:
 - a. Cl or Br
 - b. Na or Li
 - c. R or Sr
6. Electronegativity increases from left to right across the periodic table in which group:
 - a. Na or Cl
 - b. Rb or Li
 - c. R or Sr
7. Electronegativity increases from left to right across the periodic table in which group:
 - a. R or Sr
 - b. Cl or Br
 - c. Rb or Li
8. Electronegativity increases from left to right across the periodic table in which group:
 - a. R or Sr
 - b. Cl or Br
 - c. Rb or Li
9. Which is the most electronegative of the elements in the periodic table? (F is the most electronegative)
 - a. F
 - b.

Periodic Trends Questions And Answers

S Ashworth



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CliffsNotes AP Chemistry Bobrow Test Preparation Services,2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book

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