

# Periodic Trends – Atomic Mass

- The **atomic mass** describes the mass of each individual element.
- **Atomic mass** increases moving **down** each group and **right** across a period.

**Instructions:** In the periodic table below, draw two arrows representing the direction that atomic mass **increases** across groups and periods.

|   |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |
|---|---|---|---|---|---|---|---|---|---|----|----|----|--|----|----|----|----|----|--|
|   | 1 |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    | 18 |  |
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| 3 |   |   | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 |  |    |    |    |    |    |  |
| 4 |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |
| 5 |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |
| 6 |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |
| 7 |   |   |   |   |   |   |   |   |   |    |    |    |  |    |    |    |    |    |  |

**Instructions:** In each question, identify the element with the **greatest** atomic mass:

- 1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F) Iodine
- 2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B) Oxygen
- 3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe) Xenon

**Instructions:** In each question, list the elements from **least** to **greatest** atomic mass:

- 4) Magnesium (Mg), Beryllium (Be), Barium (Ba) Be, Mg, Ba
- 5) Gallium (Ga), Selenium (Se), Potassium (K) K, Ga, Se

# Periodic Trends In Atomic Properties Answers

**M Carnoy**



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