

# The Periodic Law

## SECTION 1

Select the correct answer to the following questions for the given properties.

1. \_\_\_\_\_ is the number of protons in the nucleus of an atom.
  - (a) something on the periodic table
  - (b) something on Mendeleev's original table
  - (c) something on Mendeleev's original table
  - (d) equal to the atomic number
2. \_\_\_\_\_ Mendeleev stated that elements with similar physical and chemical properties
  - (a) appeared at regular intervals in his periodic table
  - (b) appeared in order of increasing atomic weight
  - (c) appeared in order of increasing atomic weight
  - (d) appeared in order of increasing atomic weight
3. \_\_\_\_\_ The modern periodic law states that
  - (a) all elements with the same properties are placed in the same group
  - (b) the physical and chemical properties of elements are periodic functions of their atomic numbers
  - (c) elements with similar properties have similar atomic numbers
  - (d) the chemical properties of elements are periodic functions of their atomic numbers
4. \_\_\_\_\_ The discovery of the noble gases changed Mendeleev's periodic table
  - (a) adding a new
  - (b) adding a new
  - (c) adding a new
  - (d) adding a new
5. \_\_\_\_\_ The noble gases have properties of the noble gases is that they are
  - (a) inert
  - (b) unreactive
  - (c) highly reactive
  - (d) highly reactive
6. \_\_\_\_\_ Lanthan, the first element in Group 3, has an atomic number of 57. The second element in the group has an atomic number of
  - (a) 58
  - (b) 59
  - (c) 60
  - (d) 61
7. How many elements are in Group 1?
  - (a) 1
  - (b) 2
  - (c) 3
  - (d) 4
8. How many elements are in Group 2?
  - (a) 1
  - (b) 2
  - (c) 3
  - (d) 4
9. How many elements are in Group 3?
  - (a) 1
  - (b) 2
  - (c) 3
  - (d) 4

# Periodic Law Chapter 5 Test Answers

**Martha J. Gilleland, Rebecca Williams**



## **Periodic Law Chapter 5 Test Answers:**

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