

By: LEAH4SCI.COM

istry tutorial videos, ch

CCCCO $\xrightarrow{\text{PCC}}$ CCCC=O

CCCCO $\xrightarrow{\text{Strong Ox agents}}$ CCCC(=O)O

Jones Test

$\text{PCC} = \text{CrO}_3 \cdot 2\text{Pyr} \cdot \text{Cl}$

Strong ox agents

CrO_3
 $\xrightarrow{\text{H}_2\text{SO}_4}$
 $\text{Na}_2\text{Cr}_2\text{O}_7$
 $\xrightarrow{\text{H}_2\text{SO}_4}$

H_2CrO_4
 $\xrightarrow{\text{K}_2\text{C}_2\text{O}_7}$
 $\text{K}_2\text{Cr}_2\text{O}_7$
 $\xrightarrow{\text{H}_2\text{SO}_4}$
 KMnO_4



CC(C)=CC $\xrightarrow[\text{cold, dilute}]{\text{OsO}_4, \text{KMnO}_4}$ CC(C)(O)CC(O)C





only 1° & 2°
get oxidized
3° = N/A

Alkene Reduction



$\xrightarrow[\text{NH}_3(l)]{\text{Na or Li}}$  Trans
 $\xrightarrow[\text{Lindlar catalyst}]{\text{H}_2}$  cis
 $\xrightarrow[\text{Pt, Pd, ...}]{\text{H}_2}$ 



CC(=O)O $\xrightarrow{LiAlH_4}$ CCO
CC(=O)H $\xrightarrow{LiAlH_4 \text{ or } NaBH_4 \text{ or DIBALH}}$ CCO
CC(=O)C $\xrightarrow{LiAlH_4 \text{ or } NaBH_4 \text{ or DIBALH}}$ CC(O)C
CC(=O)OR $\xrightarrow{LiAlH_4}$ CCO
CC(=O)OR $\xrightarrow{1. \text{ DIBALH}}$ CC(=O)H $\xrightarrow{2. \text{ DIBALH}}$ CCO
CC#N $\xrightarrow{LiAlH_4 \text{ or } H_2, Pd/C}$ CCC(N)
CC(=O)N $\xrightarrow{LiAlH_4}$ CC(N)



$\text{Li}^+ + \text{H}-\overset{\text{H}}{\underset{\text{H}}{\text{Al}}}-\text{H}$
 $\text{NaOH} + \text{H}-\overset{\text{H}}{\underset{\text{H}}{\text{Al}}}-\text{H} = \text{weak}$



Oxidation Reduction Reactions Study Guide Answers

JR Anderson



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