Aqa a level physics paper 1- SUMMER 2023

What is meant by work function (2) - (ANS)Minimum energy required by an electron to escape from (1) a metal surface (1)

What is meant by ionisation energy? (2) - (ANS)Minimum energy required to remove an electron from an atom (1) from the ground state (1)

Why do emitted electrons have a kinetic energy that varies up to a maximum value? (4) - (ANS)Energy of a photon is fixed/constant (1)

Energy is required for electrons to overcome work function o (1)

Electrons deeper into/below the surface of the metal require more energy to overcome work function and therefore have less KE(1)

Maximum KE = Energy of photon - work function o (1)

Explain how (bright ring) diagram is evidence of the wave-particle duality of electrons? (ANS)Diffraction pattern maxima/minima (bright and dark areas) (1)
Diffraction is a property of waves- proves wave-behaviour of electrons (1)

What is meant by the ground state? - (ANS)The lowest energy state of an atom

Explain the difference between excitation and ionisation (2) - (ANS)An electron receives a discrete amount of energy for BOTH (1) Excitation promotes an electron to a higher energy level

Ionisation is when an electron receives enough energy to leave the atom

Explain why only photons of certain frequencies cause excitation (4) - (ANS)discrete energy levels (1)

Electrons need to absorb a DISCRETE amount of energy to move to a higher energy level (1).

Photons need to have a certain frequency to provide this energy, because E=hf(1) interaction is a 1:1 reaction all of the photon's energy is absorbed (1).

Explain how excitation takes place in a fluorescent tube - (ANS)Electrons flow through the tube to the applied PD causing a current to flow (1) Electrons collide with mercury atoms in the tube (1)

This raises electrons in the mercury atom to a higher energy level (1)

What is the purpose of coating in a fluorescent tube? - (ANS)UV/ High energy photons emitted from mercury atoms when the electrons in the atom de-excite (1) High energy photons absorbed by the coating (1) Coating emits photons in the visible spectrum (1)

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- 1. Understanding the eBook Sca2hp Science Paper Unit January
 - The Rise of Digital Reading Sca2hp Science Paper Unit January
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Sca2hp Science Paper Unit January
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sca2hp Science Paper Unit January
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Sca2hp Science Paper Unit January

- Personalized Recommendations
- Sca2hp Science Paper Unit January User Reviews and Ratings
- Sca2hp Science Paper Unit January and Bestseller Lists
- 5. Accessing Sca2hp Science Paper Unit January Free and Paid eBooks
 - Sca2hp Science Paper Unit January Public Domain eBooks
 - Sca2hp Science Paper Unit January eBook Subscription Services
 - Sca2hp Science Paper Unit January Budget-Friendly Options
- 6. Navigating Sca2hp Science Paper Unit January eBook Formats
 - o ePub, PDF, MOBI, and More
 - Sca2hp Science Paper Unit January Compatibility with Devices
 - Sca2hp Science Paper Unit January Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sca2hp Science Paper Unit January
 - Highlighting and Note-Taking Sca2hp Science Paper Unit January
 - Interactive Elements Sca2hp Science Paper Unit January
- 8. Staying Engaged with Sca2hp Science Paper Unit January
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sca2hp Science Paper Unit January
- 9. Balancing eBooks and Physical Books Sca2hp Science Paper Unit January
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sca2hp Science Paper Unit January
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Sca2hp Science Paper Unit January
 - Setting Reading Goals Sca2hp Science Paper Unit January
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sca2hp Science Paper Unit January

- Fact-Checking eBook Content of Sca2hp Science Paper Unit January
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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