Chapter 10 - Chemical Quantities

10.1 The Mole: A Measurement of Matter:

Mole: Otank Green's vides)

Unit of measurement

Representative Particles:

- Smallest unit of a substance (atom, molecule, formula unit, ions, anything...)
- Avogadro's number = 6.02 x 10²³ representative particles (RP)
- 6.02 x 10²³ representative particles= 1 mol

Molar Mass:

- Molar mass= g/mol
- · Molar mass of:

-an element = atomic mass (g)

Ex. 24.3 g/mol = molar mass of Mg

-a compound = sum of the atomic masses (g) of each element

Ex. 58.44g/mol = molar mass of NaCl

Molar mass= 1 mol

Molar Volume:

- Molar volume= 22.4 L/mol
- Standard temperature and pressure (STP) = temp= 0° C pressure = 1 atm
- 22.4 L = 1 mol (at STP)

Calculating Molar mass from Density:

 molar mass (g/mole) = density at STP (g/L) x molar volume at STP (22.4L/mole)

Calculating Density from Molar mass:

 Density at STP (g/L) = molar mass (g/mole) x molar volume at STP (1mole/22.4L)

Density (g/L) =
$$\begin{pmatrix} grams \\ 1 \text{ mole} \end{pmatrix} \begin{pmatrix} 1 \text{ mole} \\ 22.4 \text{ L} \end{pmatrix}$$

<u>Prentice Hall Chemistry Chemical Quantities Chapter 10</u> <u>Lecture Notes</u>

Rosina Ehmann

Prentice Hall Chemistry Chemical Quantities Chapter 10 Lecture Notes:

The Enigmatic Realm of **Prentice Hall Chemistry Chemical Quantities Chapter 10 Lecture Notes**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Prentice Hall Chemistry Chemical Quantities Chapter 10 Lecture Notes** a literary masterpiece penned by a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of people who partake in its reading experience.

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