

Preliminary Chemistry Notes - 2020

Glossary of key terms:

Decomposition - The process by which organic substances are broken down into simpler organic matter

Lustre – How shiny a substance appears and relates to how well the substance is able to reflect light; in general, metals are lustrous, and non-metals are dull

Matter – The material substance that makes up the physical universe

Molecule - A group of two or more atoms covalently bonded together (usually refers to non-metals)

- Molecules can also be defined as particles which move independently of each other
- Thus, noble gases (group 8) which are chemically stable usually exists as single particles moving independently aka monoatomic molecules (containing one atom)
- Diatomic molecules are molecules with two atoms, e.g. H_2 , Cl_2

Compound - Molecules that have at least two elements different, chemically combined in a specific ratio. They are represented by a chemical formula

Mixture – A substance of variable composition, containing two or more elements or compounds that are not chemically bonded together

Covalent Bond – A covalent bond, also called a molecular bond, is a chemical bond that involves the sharing of electron pairs between atoms

Period - A period in the periodic table is a row of chemical elements. All elements in a row have the same number of electron shells. Each next element in a period has one more proton and is less metallic than its predecessor

Group - The periodic table also has a special name for its vertical columns. Each column is called a group. The elements in each group have the same number of electrons in the outer orbital. Those outer electrons are also called valence electrons

Ion – Atom or molecule with a net electrical charge

Polyatomic Ion - A polyatomic ion, also known as a molecular ion, is a **charged** chemical species composed of two or more atoms covalently bonded

Solution – A homogeneous mixture of two or more substances, with a minor substance being uniformly distributed in a major substance. The minor substance is called the solute and the major substance is called the solvent. An example is saltwater (where salt = solute, and water = solvent). Also, the solute is dissolved in the solvent, and solute particles cannot be seen under a light microscope

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WJ Hussar



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