Predicting Reaction Products

Predict the products for the following reactions, balance the equation, then classify the type of reactions:

- ___Na+__FeBr₃→ 1)
- ____ NaOH + ____ H₂SO4 → 2)
- 3) C₂H₄O₂ + O₂ →
- NH₃ + H₂O → 40
- 5) PbSO. + AgNO₃ →
- 60 PBr_b →
- 7) HBr+ Fe→
- 8) $_{\text{L}}$ KMnO₄ + $_{\text{L}}$ ZnCl₂ \rightarrow
- 99 $MnO_2 + Sn(OH)_4 \rightarrow$
- O₂ + C₂H₂₂O₂ → 10)
- 11) H₂O₂ →
- 12) PtCl₄ + Cl₂ →

Predicting Reaction Products Answers

David R. Klein, Laurie S. Starkey

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textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve **General Chemistry Workbook** Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of problems solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and guestions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material Shriver and Atkins' Inorganic Chemistry Peter Atkins, 2010 Inorganic Chemistry fifth edition represents an integral part of a student's chemistry education Basic chemical principles are set out clearly in Foundations and are fully developed throughout the text culminating in the cutting edge research topics of the Frontiers which illustrate the dynamic nature of inorganic chemistry

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