

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $C_6H_{12}O_6$ (molecular)

Formula represents the number of _____ of each
in a single molecule.

Empirical $C_6H_{12}O_6$ (molecular)

Formula represents the proportions of _____ of
positive and negative ions in a formula
unit.

Atomic Mass

The proportions that atoms combine through
ions together.

Isotopes

$^{12}_6C$ isotope isotope isotope

$^{13}_6C$ isotope isotope isotope

Subscript

$^{12}_6C$ isotope isotope

$^{13}_6C$ isotope isotope isotope

$^{14}_6C$ isotope

Note Taking Guide Physics 601

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