

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ relationship _____ equated _____

2. $40g = 40r + 4 \cdot 20g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ _____ from the _____ reaction in the _____

3. Problem

When aluminum chloride reacts with aluminum hydroxide, aluminum chloride is produced



moles of

$AlCl_3$	$Al(OH)_3$	HCl
2	3	6
$AlCl_3$	$Al(OH)_3$	HCl

2. If you start with 4 moles of $AlCl_3$, how many moles of $Al(OH)_3$ will be produced?

2 moles $AlCl_3$ → 3 moles $Al(OH)_3$ and 6 moles HCl → 4 moles $Al(OH)_3$

Note Taking Guide Episode 801 Stoichiometry Answers

Anthony S. Fauci



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