

Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON-METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.) H_2 hydrogen is diatomic



Single Bond

2.) F_2 fluorine is diatomic



3.) O_2 oxygen is diatomic



Double Bond

4.) N_2 nitrogen is diatomic. Is this a triple bond?



5.) BF_3 you need 3 fluorine atoms here



6.) Ammonia NH_3 hint: how many hydrogen atoms are needed?



7.) Carbon dioxide CO_2



8.) Methane CH_4 careful here 4 hydrogen atoms needed



9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!



10.) SO_2 hint: one pair of electrons from sulfur must be slit up for this one to work.



Prentice Hall Chemistry Covalent Bonding Answers

Michael B. Smith



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Prentice Hall Chemistry, 2000 Fundamentals of Materials Engineering- A Basic Guide Shashanka

Rajendrachari, Orhan Uzun, 2021-02-22 Fundamentals of Materials Engineering A Basic Guide is a helpful textbook for readers learning the basics of materials science This book covers important topics and fundamental concepts of materials engineering including crystal structure imperfections mechanical properties of materials polymers powder metallurgy corrosion and composites The authors have explained the concepts in an effective way and by using simple language for the benefit of a broad range of readers This book is also beneficial to the students in engineering courses at B Sc M Sc and M Tech levels **Computational Chemistry** Errol G. Lewars, 2010-11-09 This corrected second edition contains new material

which includes solvent effects the treatment of singlet diradicals and the fundamentals of computational chemistry Computational Chemistry Introduction to the Theory and Applications of Molecular and Quantum Mechanics is an invaluable tool for teaching and researchers alike The book provides an overview of the field explains the basic underlying theory at a meaningful level that is not beyond beginners and it gives numerous comparisons of different methods with one another and with experiment The following concepts are illustrated and their possibilities and limitations are given potential energy surfaces simple and extended Hückel methods ab initio AM1 and related semiempirical methods density functional theory DFT Topics are placed in a historical context adding interest to them and removing much of their apparently arbitrary aspect The large number of references to all significant topics mentioned should make this book useful not only to undergraduates but also to graduate students and academic and industrial researchers **Addison Wesley Chemistry 5th Edition**

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Madrazo, Jr., 1990 *Prentice Hall Science Explorer: Chemical Interactions* Pearson/Prentice Hall, Michael J. Padilla, 2004-10

Chemistry Theodore P. Perros, 1967 *Prentice Hall Scientific Learning System* Prentice-Hall Staff, 1994 **March's**

Advanced Organic Chemistry Michael B. Smith, 2020-02-19 The completely revised and updated definitive resource for students and professionals in organic chemistry The revised and updated 8th edition of March's Advanced Organic Chemistry Reactions Mechanisms and Structure explains the theories of organic chemistry with examples and reactions This book is the

most comprehensive resource about organic chemistry available Readers are guided on the planning and execution of multi step synthetic reactions with detailed descriptions of all the reactions The opening chapters of March s Advanced Organic Chemistry 8th Edition deal with the structure of organic compounds and discuss important organic chemistry bonds fundamental principles of conformation and stereochemistry of organic molecules and reactive intermediates in organic chemistry Further coverage concerns general principles of mechanism in organic chemistry including acids and bases photochemistry sonochemistry and microwave irradiation The relationship between structure and reactivity is also covered The final chapters cover the nature and scope of organic reactions and their mechanisms This edition Provides revised examples and citations that reflect advances in areas of organic chemistry published between 2011 and 2017 Includes appendices on the literature of organic chemistry and the classification of reactions according to the compounds prepared Instructs the reader on preparing and conducting multi step synthetic reactions and provides complete descriptions of each reaction The 8th edition of March s Advanced Organic Chemistry proves once again that it is a must have desktop reference and textbook for every student and professional working in organic chemistry or related fields Winner of the Textbook Academic Authors Association 2021 McGuffey Longevity Award

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MATERIALS SCIENCE AND ENGINEERING : PROBLEMS WITH SOLUTIONS SHETTY, M.N., 2015-12-01 This book with analytical solutions to 260 select problems is primarily designed for the second year core course on materials science The treatment of the book reflects the author s experience of teaching this course comprehensively at IIT Kanpur for a number of years to the students of engineering and 5 year integrated disciplines The problems have been categorised into five sections covering a wide range of solid state properties Section 1 deals with the dual representation of a wave and a particle and then comprehensively explains the behaviour of particles within potential barriers It provides solutions to the problems that how the energy levels of a free atom lead to the formation of energy bands in solids The statistics of the distribution of particles in different energy states in a solid has been detailed leading to the derivation of Maxwell Boltzmann Bose Einstein and Fermi Dirac statistics and their mutual relationships Quantitative derivation of the Fermi energy has been obtained by considering free electron energy distribution in solids and then considering Fermi Dirac distribution as a function of temperature The derivation of the Richardson s equation and the related work function has been quantitatively dealt with The phenomenon of tunnelling has been dealt with in terms of quantum mechanics whereas the band structure and electronic properties of materials are given quantitative treatment by using Fermi Dirac distribution function Section 2 deals with the nature of the chemical bonds types of bonds and their effect on properties followed by a detailed presentation of crystal structures of some common materials and a discussion on the structures of C60 and carbon nanotubes Coordination and packing in crystal structures are considered next followed by a detailed X ray analysis of simple crystal structures imperfections in crystals diffusion phase equilibria and mechanical behaviour Section 3 deals with thermal and electrical properties and their mutual relationships Calculations of Debye frequency Debye temperature and Debye specific heat are presented in great detail A brief section on superconductivity considers both the

conventional and the high TC superconductors Sections 4 and 5 deal with the magnetic and dielectric materials considering magnetic properties from the point of view of the band theory of solids Crystal structures of some common ferrites are given in detail Similarly the displacement characteristics in dielectrics are considered from their charge displacements giving rise to some degree of polarization in the materials

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