

Chemistry - Review Chapters 4-5: 6

Directions: Answer each of the following questions. You need not use complete sentences.

1. By what property did Mendeleev arrange the elements?

ATOMIC MASS

2. By what property did Moseley suggest that the periodic table be arranged?

ATOMIC NUMBER

3. What is a period? How many are there in the periodic table?

A HORIZONTAL ROW IN THE PERIODIC TABLE; 7

4. What is a group (also called a family)? How many are there in the periodic table?

A VERTICAL COLUMN IN THE PERIODIC TABLE; 18

5. State the number of valence electrons in an atom of:

a. sulfur	b. calcium	c. chlorine	d. arsenic
6	2	7	5

6. Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10	b. 18	c. 36	d. 90
Ne, NEON	Ar, ARGON	Kr, KRYPTON	Th, THORIUM

7. List, by number, both the period and group of each of these elements.

	<u>Symbol</u>	<u>Period</u>	<u>Group</u>
a. beryllium	Be	2	2
b. iron	Fe	4	8
c. lead	Pb	6	14

8. Which of the following pairs of elements belong to the same period?

a. Na and Cl	b. Na and Li	c. Na and Cu	d. Na and Ne
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9. Which of the following pairs of elements belong to the same group?

a. H and He	b. Li and Be	c. C and Pb	d. Ga and Ge
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10. How does an element's period number relate to the number of the energy level of its valence electrons?

PERIOD NUMBER = ENERGY LEVEL OF VALENCE ELECTRONS

11. What are the transition elements?

GROUPS 3-12

12. In what type of orbitals are the actinide and lanthanide electrons found?

f ORBITALS

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Wenbin Ji



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Table of Contents Prentice Hall Chemistry Answer Key Chapter 5

1. Understanding the eBook Prentice Hall Chemistry Answer Key Chapter 5
 - The Rise of Digital Reading Prentice Hall Chemistry Answer Key Chapter 5
 - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Chemistry Answer Key Chapter 5
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Chemistry Answer Key Chapter 5
 - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Chemistry Answer Key Chapter 5
 - Personalized Recommendations
 - Prentice Hall Chemistry Answer Key Chapter 5 User Reviews and Ratings
 - Prentice Hall Chemistry Answer Key Chapter 5 and Bestseller Lists
5. Accessing Prentice Hall Chemistry Answer Key Chapter 5 Free and Paid eBooks
 - Prentice Hall Chemistry Answer Key Chapter 5 Public Domain eBooks
 - Prentice Hall Chemistry Answer Key Chapter 5 eBook Subscription Services
 - Prentice Hall Chemistry Answer Key Chapter 5 Budget-Friendly Options
6. Navigating Prentice Hall Chemistry Answer Key Chapter 5 eBook Formats

- ePub, PDF, MOBI, and More
 - Prentice Hall Chemistry Answer Key Chapter 5 Compatibility with Devices
 - Prentice Hall Chemistry Answer Key Chapter 5 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Prentice Hall Chemistry Answer Key Chapter 5
 - Highlighting and Note-Taking Prentice Hall Chemistry Answer Key Chapter 5
 - Interactive Elements Prentice Hall Chemistry Answer Key Chapter 5
8. Staying Engaged with Prentice Hall Chemistry Answer Key Chapter 5
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Chemistry Answer Key Chapter 5
9. Balancing eBooks and Physical Books Prentice Hall Chemistry Answer Key Chapter 5
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Prentice Hall Chemistry Answer Key Chapter 5
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Prentice Hall Chemistry Answer Key Chapter 5
- Setting Reading Goals Prentice Hall Chemistry Answer Key Chapter 5
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Prentice Hall Chemistry Answer Key Chapter 5
- Fact-Checking eBook Content of Prentice Hall Chemistry Answer Key Chapter 5
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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